

Critical loads in Europe: overview and latest developments

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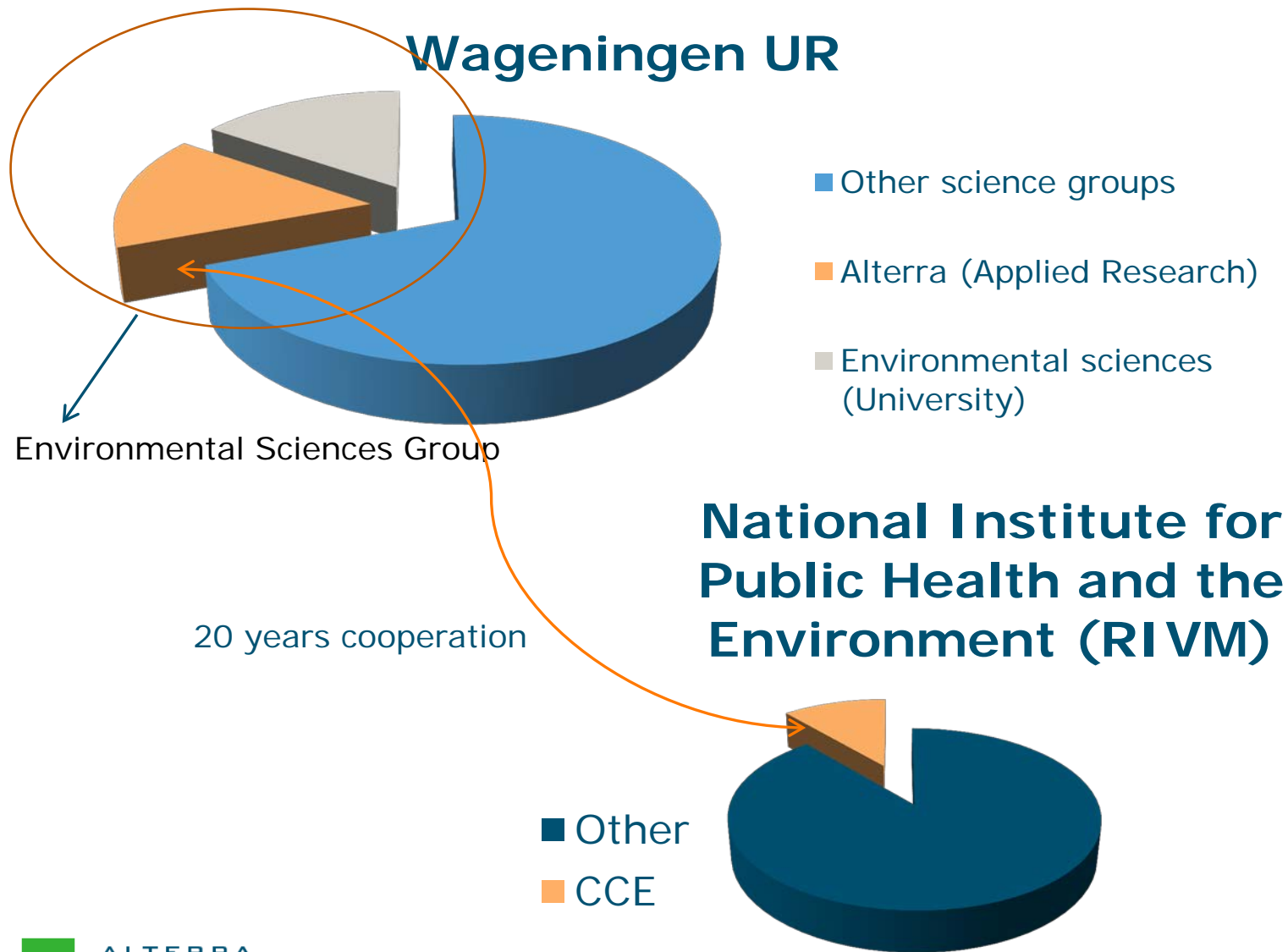
With contributions from Luc Bonten, Janet Mol, Wieger Wamelink, Wim de Vries (Alterra) and Max Posch (RIVM-CCE)



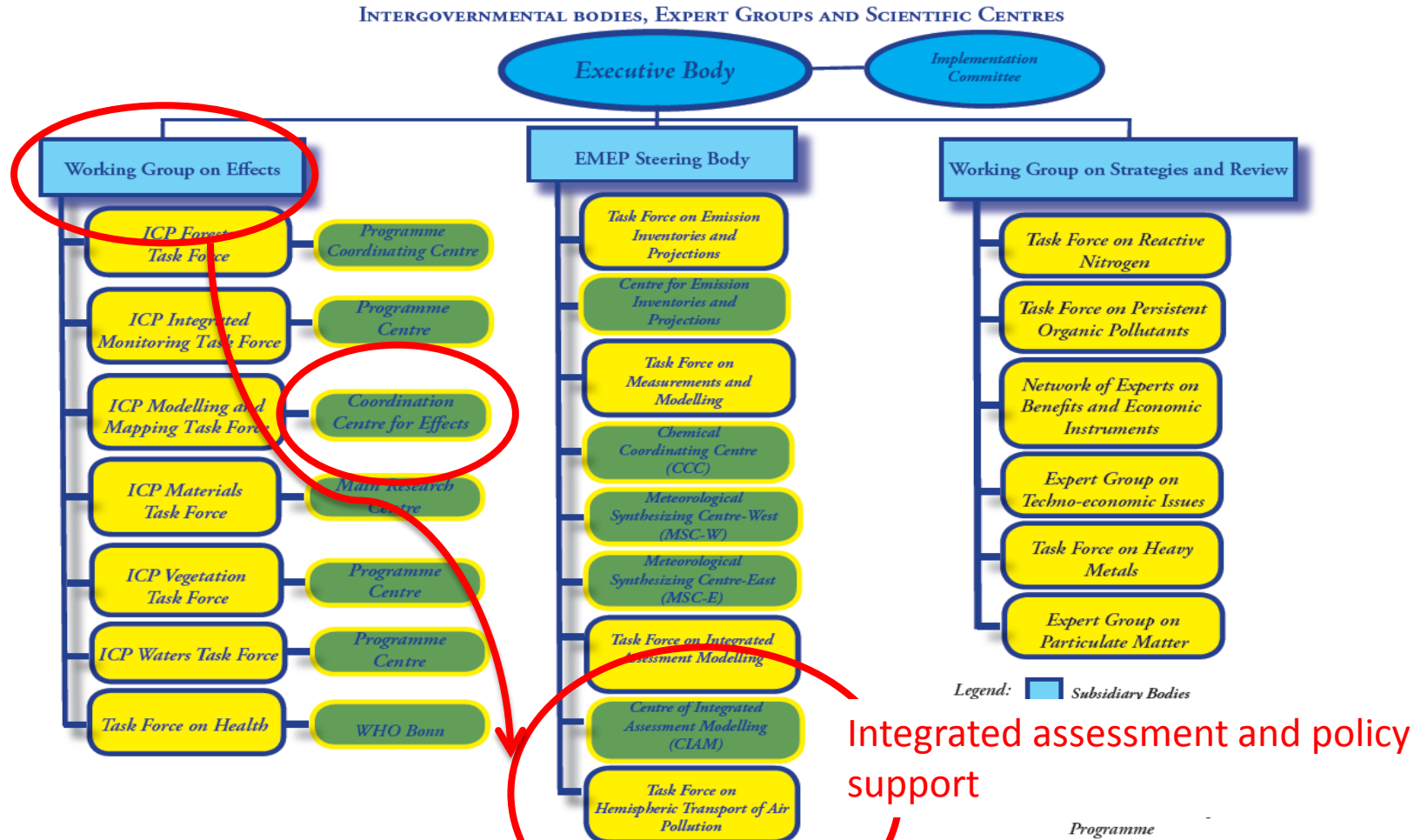
Contents

- Alterra and the CCE
- The critical load process in Europe
 - Organisation
 - Achievements
 - Critical loads for N as a nutrient: using biodiversity criteria for critical loads
- Critical loads for heavy metals
 - Methods
 - Results
- Conclusions

Alterra, Wageningen UR and RIVM-CCE

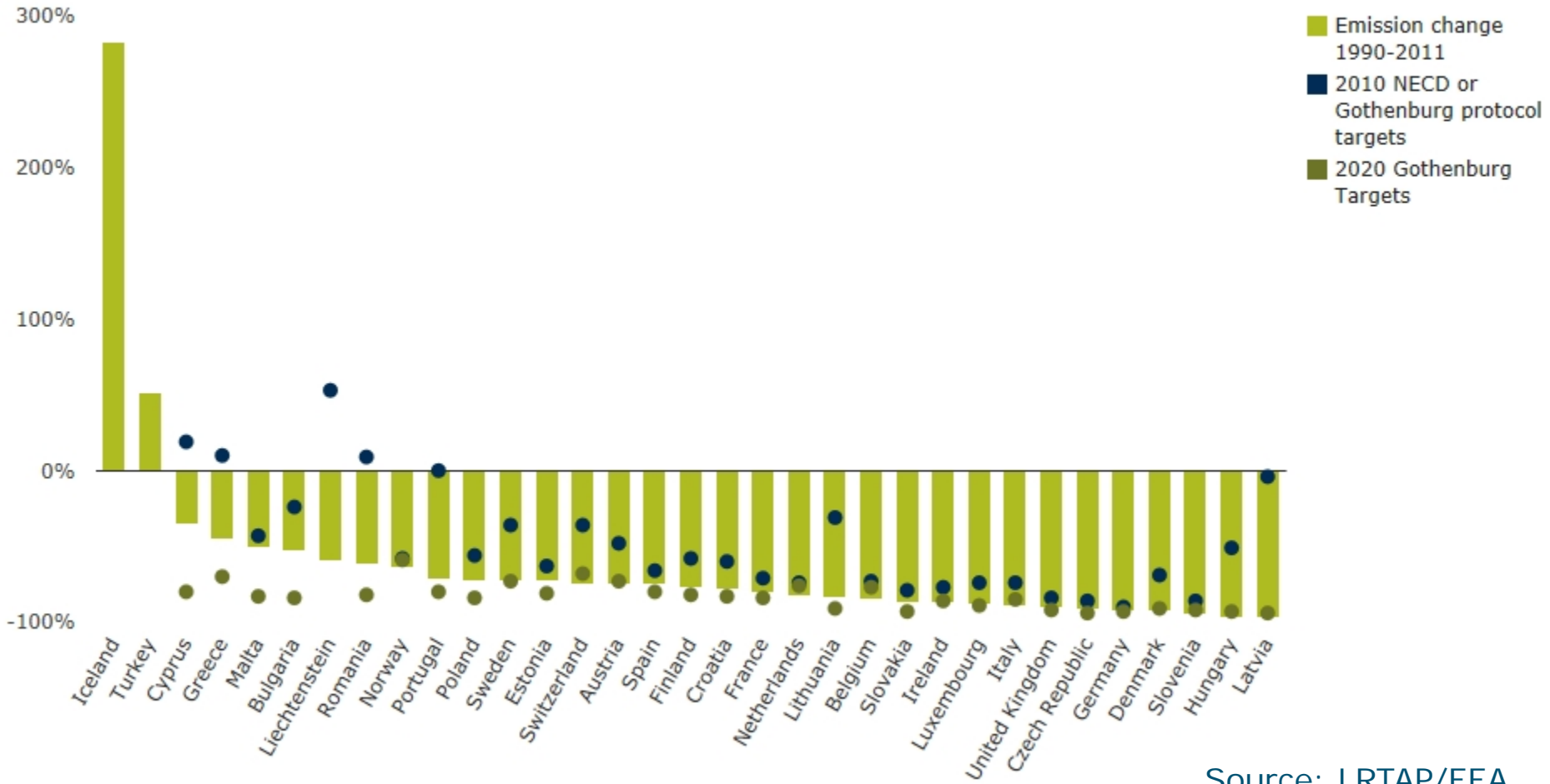


The critical load process in Europe: The 1979 Geneva Convention on Long-range Transboundary Air Pollution



Achievements: emission reductions SO_x

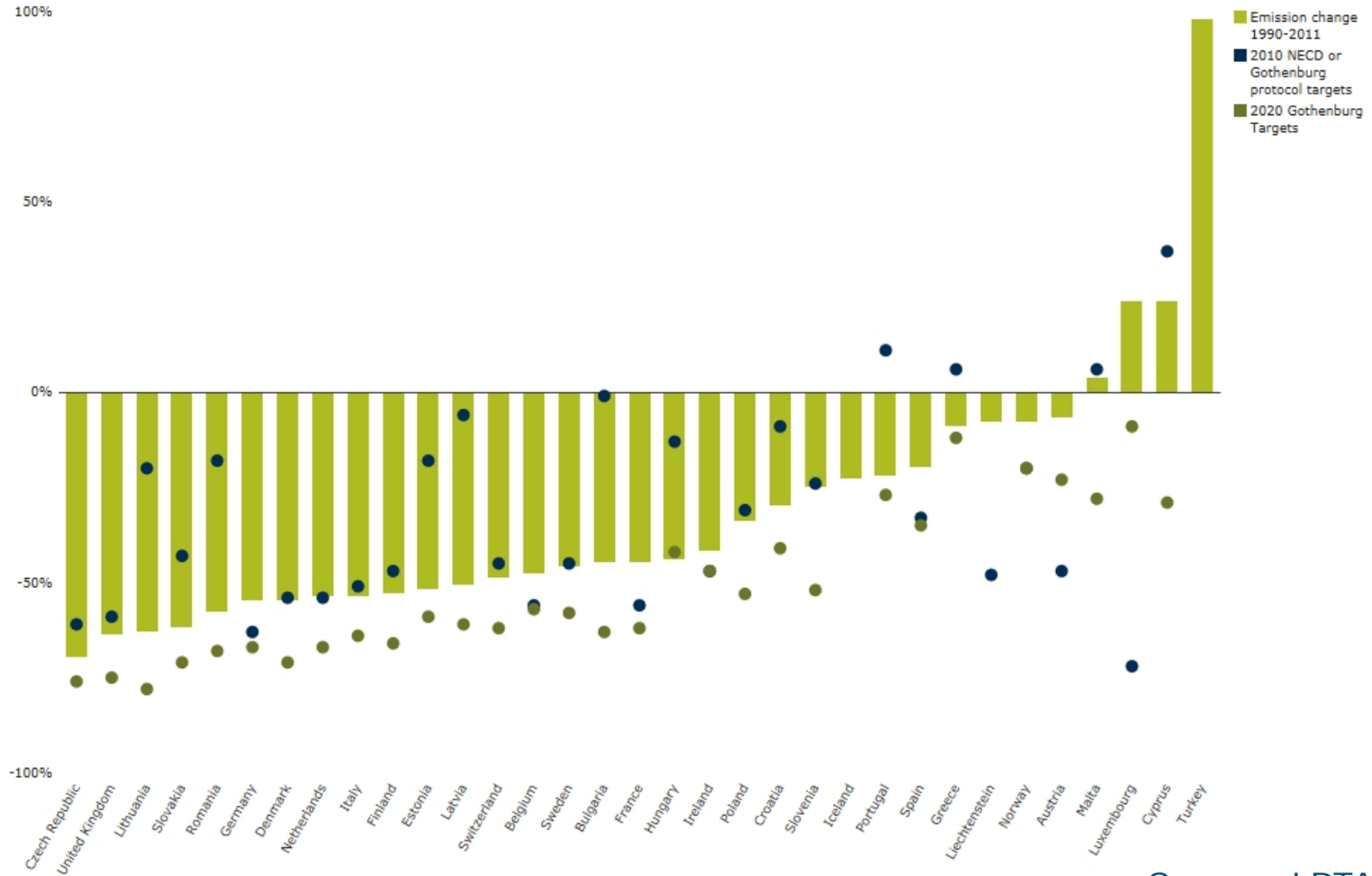
Chart – Change in emissions of sulphur oxides compared with the 2010 NECD and Gothenburg protocol targets



Source: LRTAP/EEA

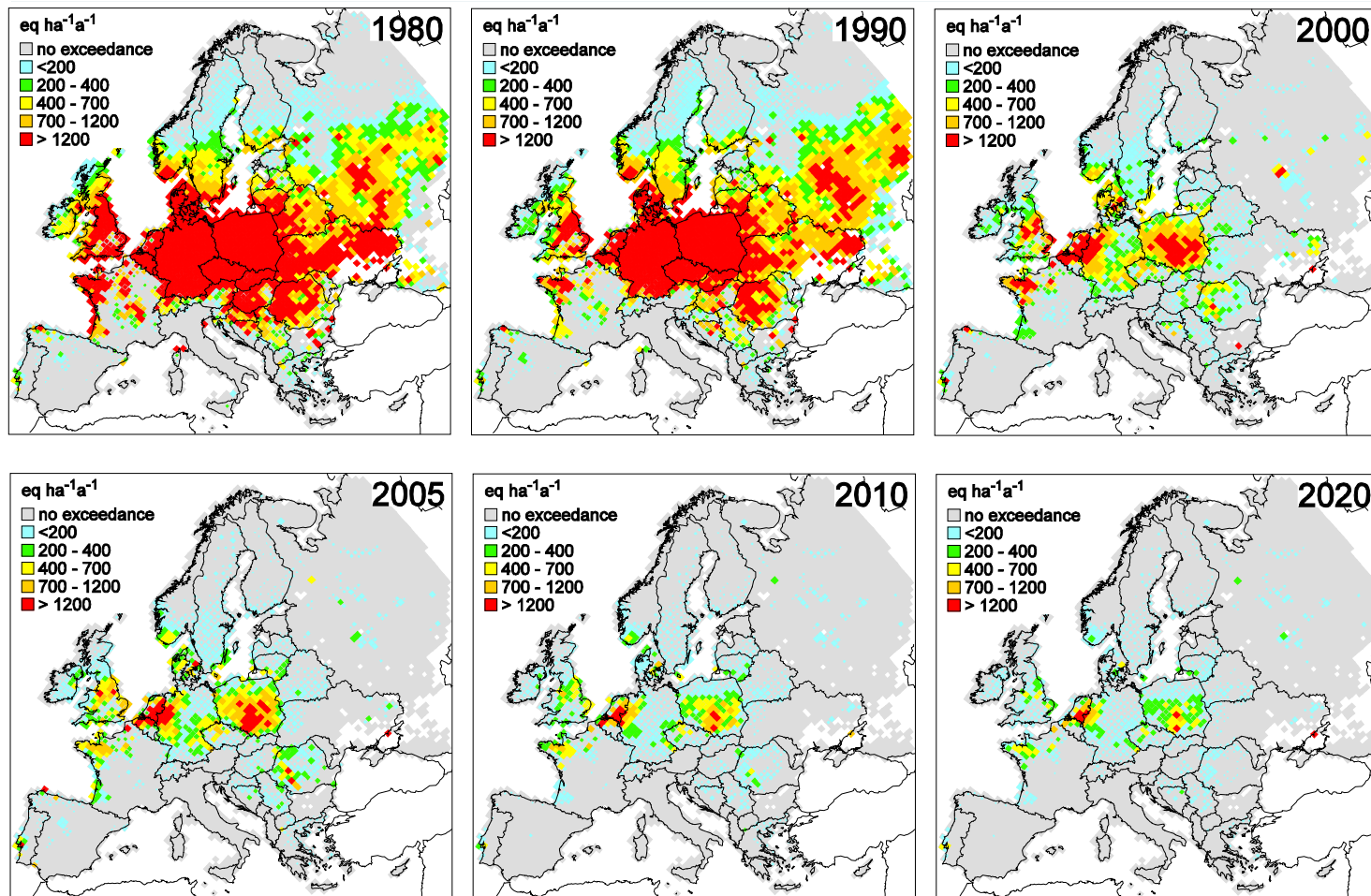
Achievements: emission reductions NO_x

Chart — Change in emissions of nitrogen oxides compared with the 2010 NECD and Gothenburg protocol targets



Source: LRTAP/EEA

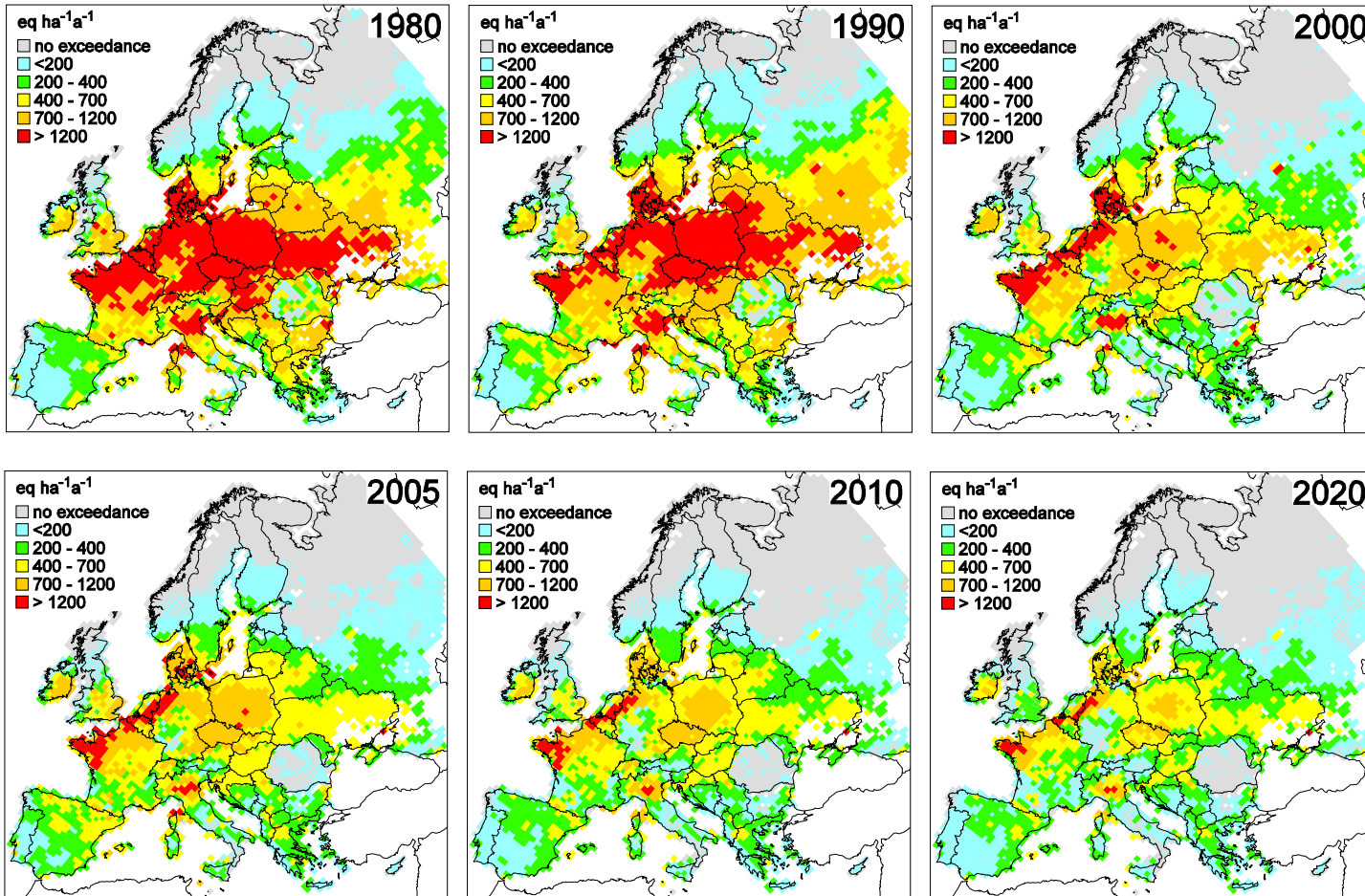
Achievements: trends in exceedances: acidity



Areas where critical loads for acidification are exceeded by acid depositions (EMEP50 model;
Revised Gothenburg Protocol (RGP))

Source: ICPM&M/CCE

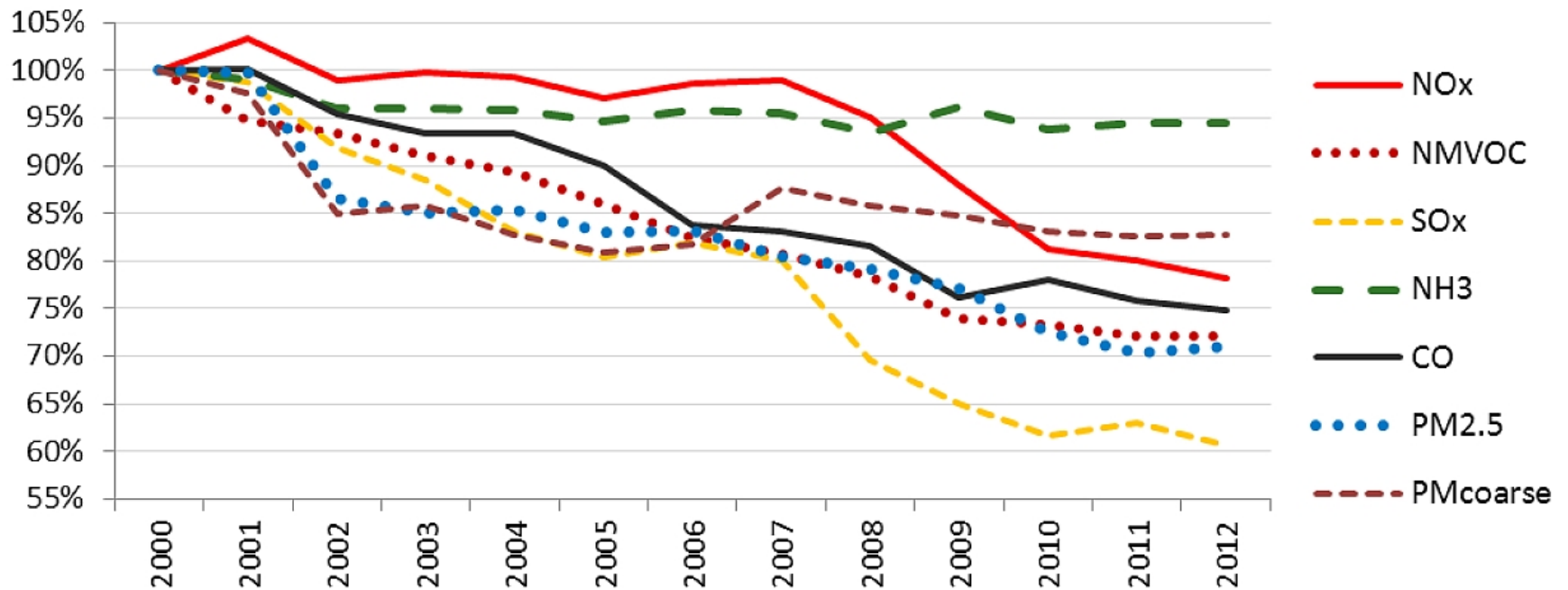
Achievements: trends in exceedances: CLnutN



Areas where critical loads for eutrophication are exceeded by nitrogen depositions (EMEP50 model; Revised Gothenburg Protocol (RGP))

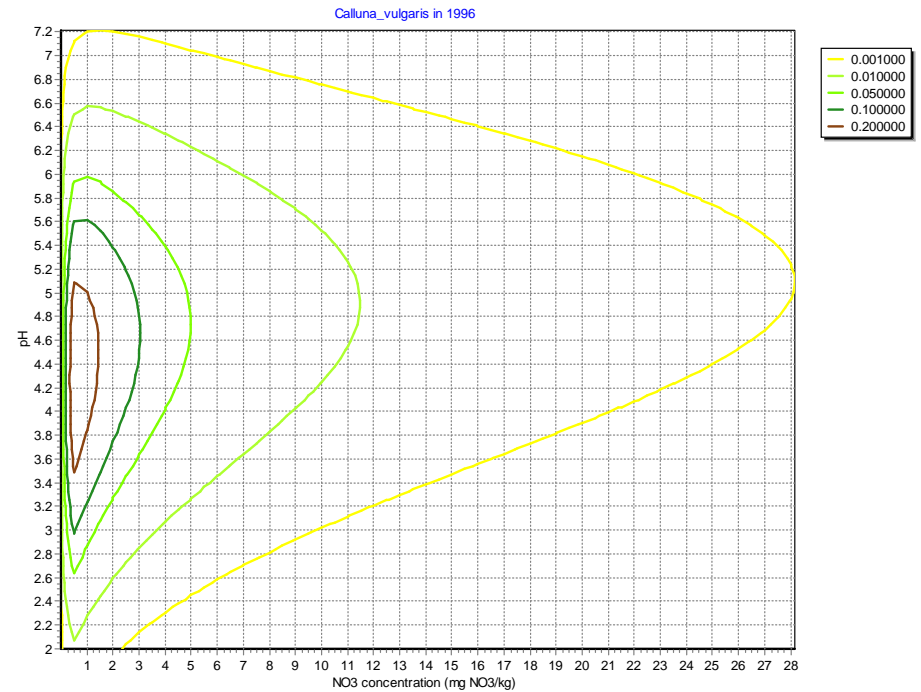
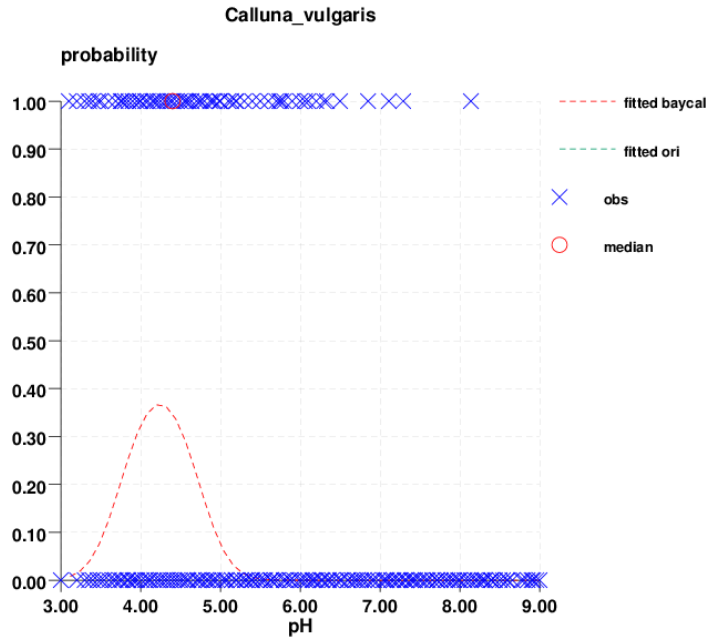
Source: ICPM&M/CCE

Emission trends

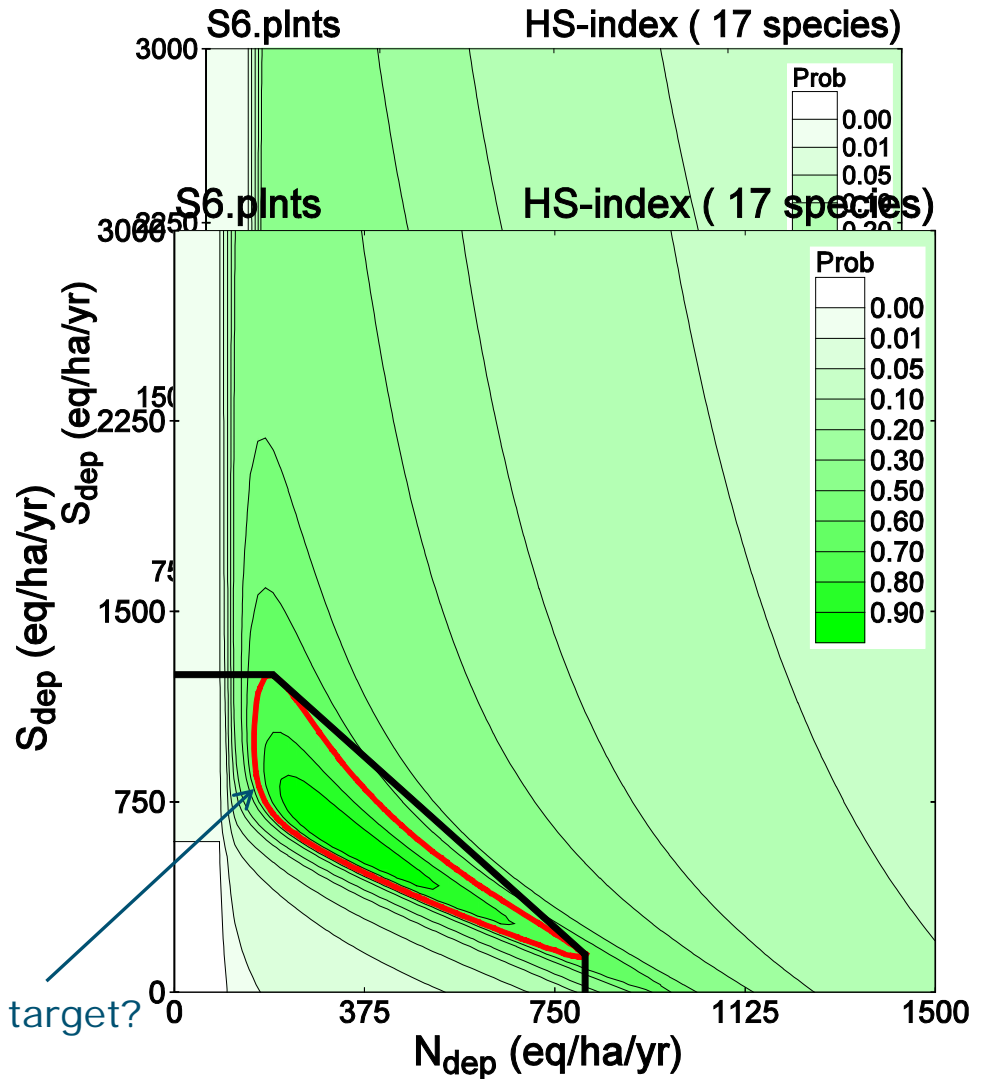
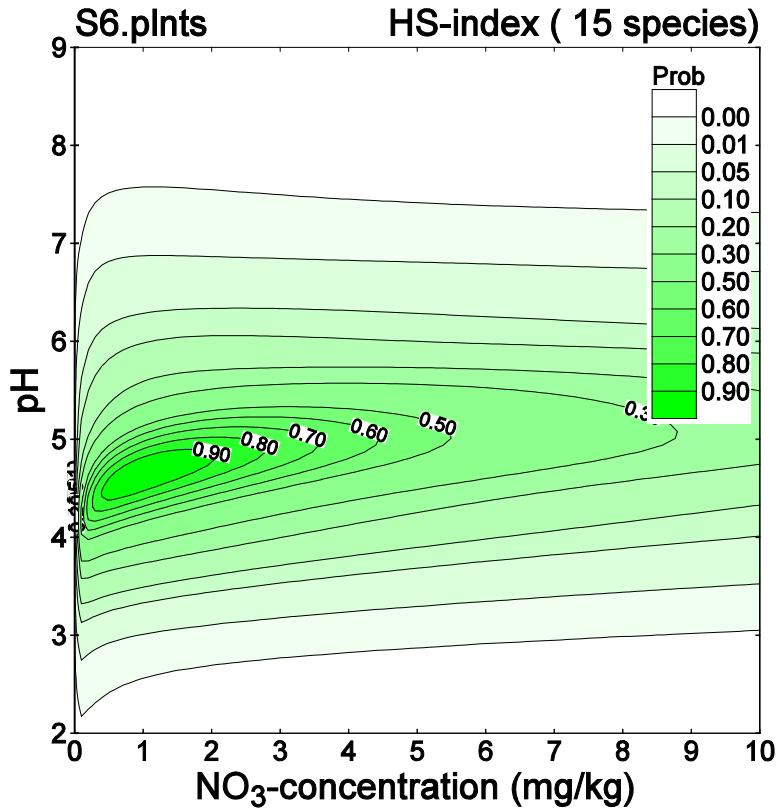


Source: EMEP REPORT 1/2014

Future directions: Critical loads based on biodiversity endpoints

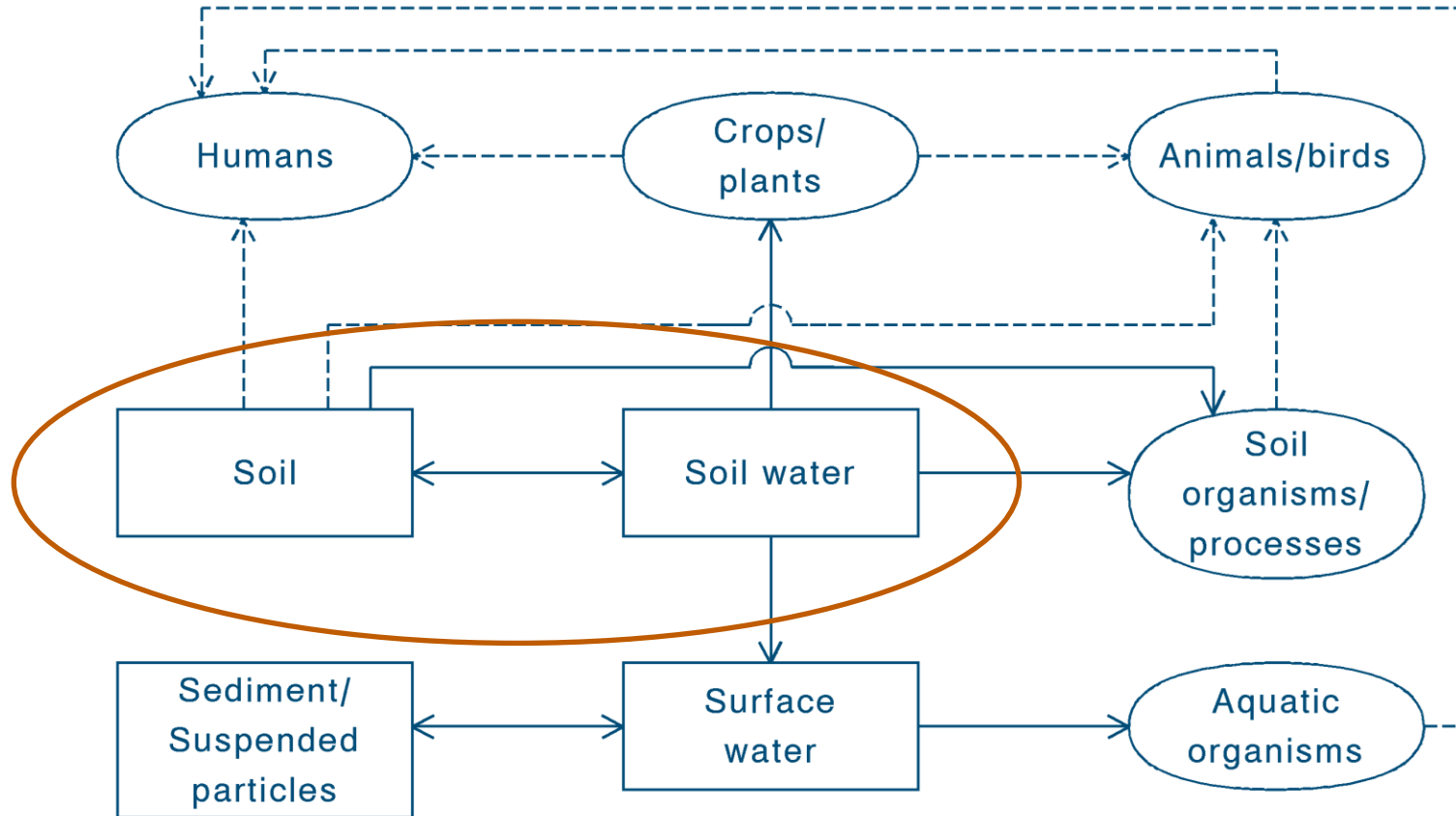


From response to critical load



Who defines the target?

Part 2: Critical loads for heavy metals



figme4c.tce

Critical load for terrestrial ecosystems: uptake + leaching:

$$CL(M) = M_u + M_{le(crit)}$$

$$M_u = f_{Mu} \cdot Y_{ha} \cdot [M]_{ha}$$

f_{Mu} = fraction of metal uptake within the considered soil layer

Y_{ha} = yield of harvestable biomass ($\text{kg} \cdot \text{ha}^{-1} \cdot \text{a}^{-1}$)

$[M]_{ha}$ = metal content in harvestable parts of the plant ($\text{g} \cdot \text{kg}^{-1}$)

$$M_{le(crit)} = C_{le} \cdot Q_{le} \cdot [M]_{tot,sw(crit)}$$

Q_{le} = leaching flux ($\text{m} \cdot \text{a}^{-1}$)

$[M]_{tot,sw(crit)}$ = critical total concentration of M in soil water ($\text{mg} \cdot \text{m}^{-3}$)

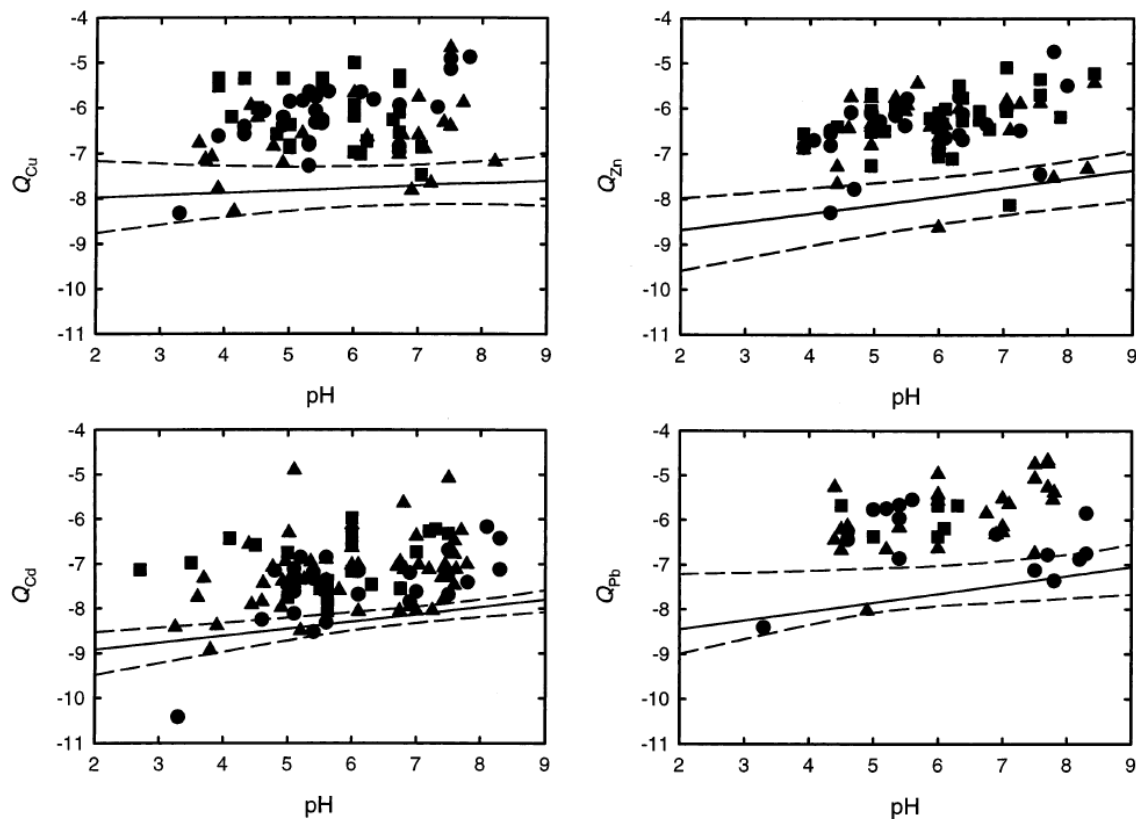
C_{le} = unit conversion factor

Which critical concentration?

- Critical metal concentrations in ground water (Cd, Pb, Hg) in view of human health effects through intake of drinking water
WHO criteria:
Pb: 10 mg.m^{-3}
Cd: 3 mg.m^{-3}
Hg: 6 mg.m^{-3}
- Critical concentrations of metal ions in soil solution (Cd, Pb) in view of ecotoxicological effects on soil microorganisms, plants and invertebrates
From toxicity data, as a function of pH and DOC
- Critical metal contents in the soil (Hg) in view of ecotoxicological effects on soil microorganisms and invertebrates in the forest humus layer
Compute Hg in solution from critical Hg in soil

From toxicity data to critical concentrations

1: Toxicity data

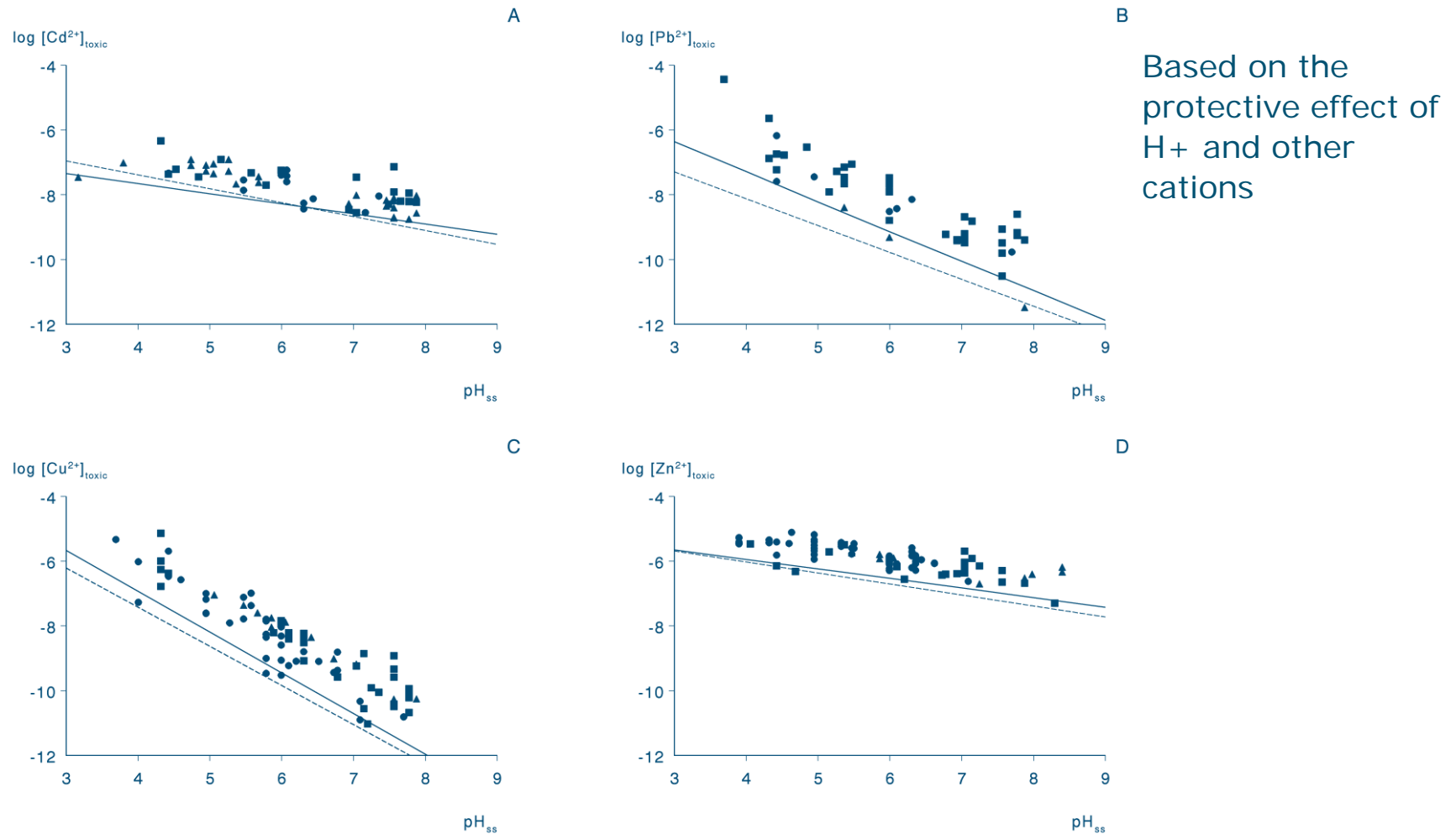


Q represents $[\log M_{\text{soil, toxic}} + [b/c] \log \text{OM}]$

FIGURE 1. Plots of Q against pH (eq 6; Table 7) for Cu, Zn, Cd, and Pb. Circles: plant data; squares: invertebrate data; and triangles: microbial process data. Solid line: median critical limit function and dotted lines: 90% confidence limits on the median critical limit.

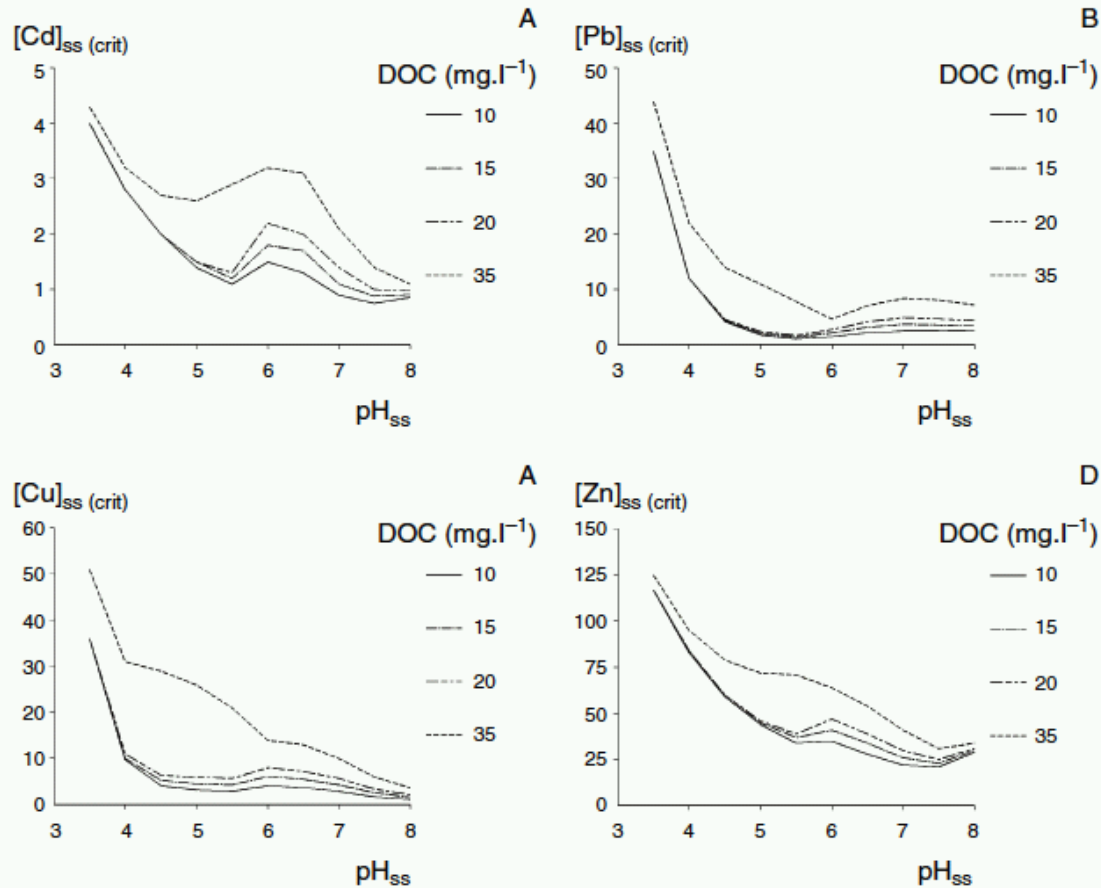
From toxicity data to critical concentrations

2: Critical limit functions



From toxicity data to critical concentrations

3: Total critical concentrations



Critical total concentrations as a function of pH and DOC using the WHAM model

Hg: two approaches

1: Using critical limit for the solid phase

$$[\text{Hg}]_{\text{dis, sdw(crit)}} = [\text{Hg}]_{\text{OM(crit)}} \cdot f_f \cdot [\text{DOM}]_{\text{sdw}} \cdot \text{csdw}$$

$[\text{Hg}]_{\text{dis, sdw(crit)}}$ = critical total Hg concentration in soil drainage water (mg m^{-3})

$[\text{Hg}]_{\text{OM(crit)}}$ = critical limit for Hg concentration in solid organic matter (OM) ($0.5 \text{ mg (kg OM)}^{-1}$).

f_f = fractionation ratio, describing the Hg on organic matter in solution (DOM) relative to that in solids (OM) (-),

$[\text{DOM}]_{\text{sdw}}$ = concentration of dissolved organic matter in soil drainage water (g m^{-3}),

csdw = $10^{-3} \text{ kg g}^{-1}$, factor for appropriate conversion of mass units.

This is the approach given in the mapping manual
(but is pH independent)

Hg: two approaches

2: Using critical free concentrations

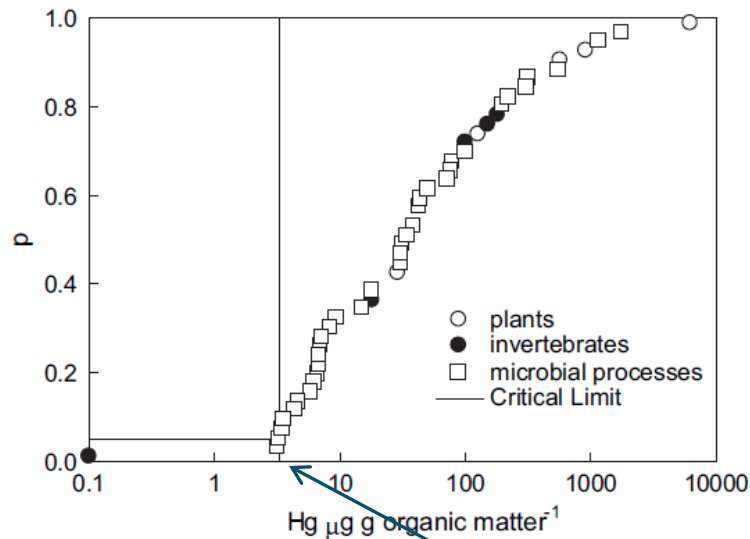


Fig. 1. Species sensitivity distributions for Hg(II) chronic toxicity end-points in soil, expressed in terms of soil solids (upper panel) and soil organic matter (lower panel). The derivation of Critical Limits at the 5th percentile is demonstrated with the horizontal and vertical lines.

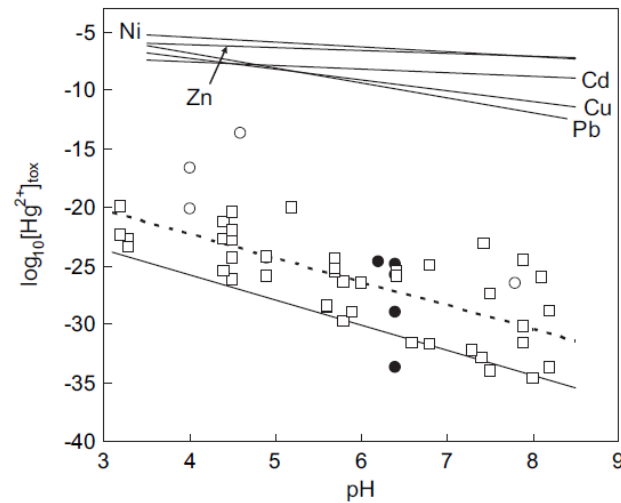
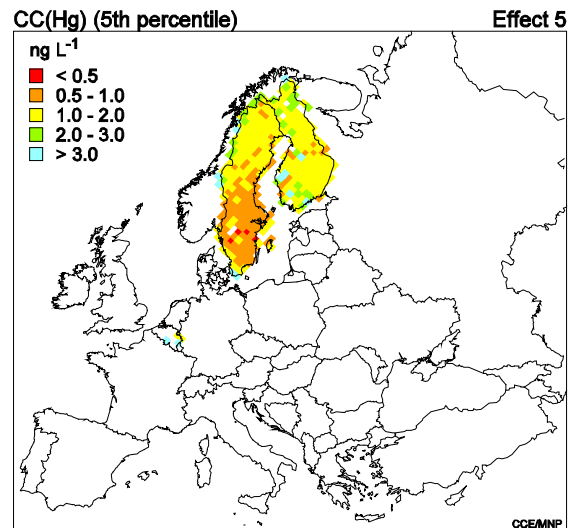
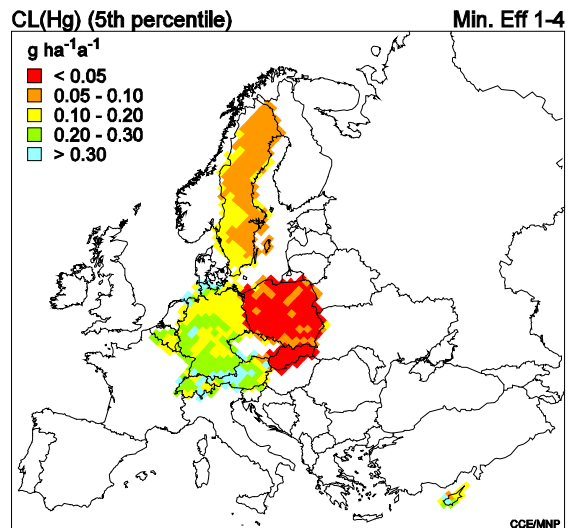
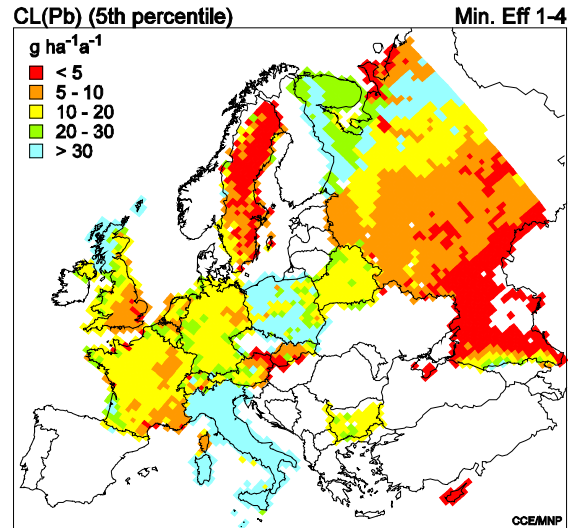
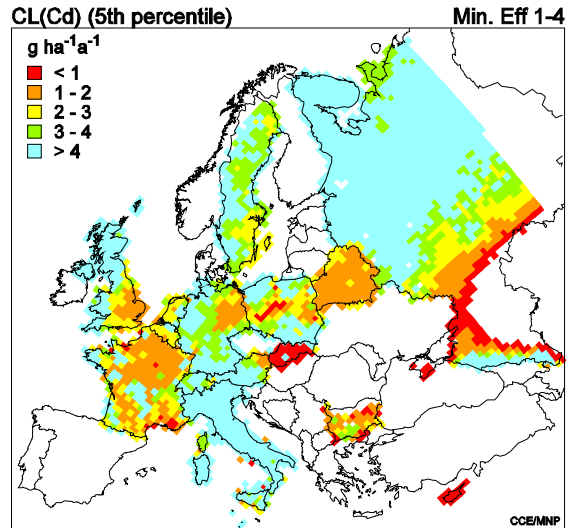


Fig. 2. Toxic end-points expressed as $[Hg^{2+}]_{tox}$, symbols as for Fig. 1. The dashed line is the median regression, and the solid line the derived Critical Limit Function (CLF). CLFs for Ni, Cu, Zn, Cd and Pb are also shown.

Critical limit about 3.3 mg.kg^{-1}
(manual: 0.5 mg.kg^{-1})

Some results: call for data (2004/2005)

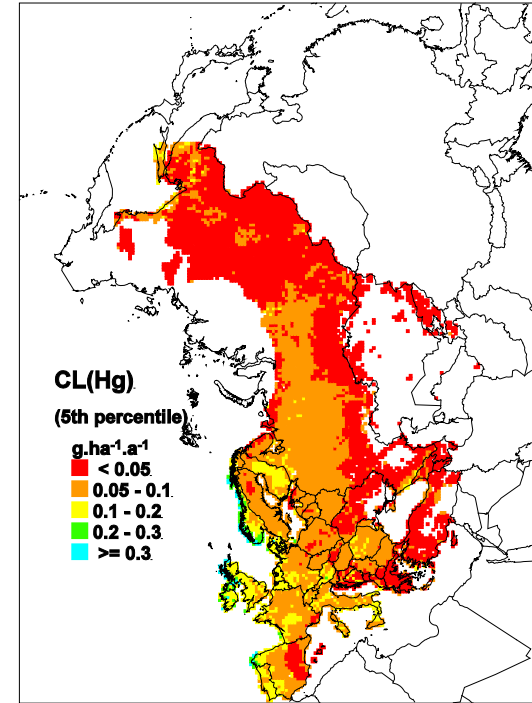
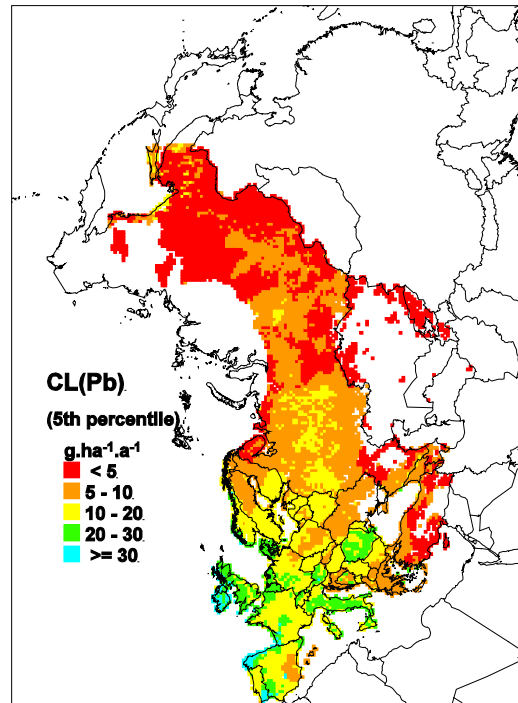
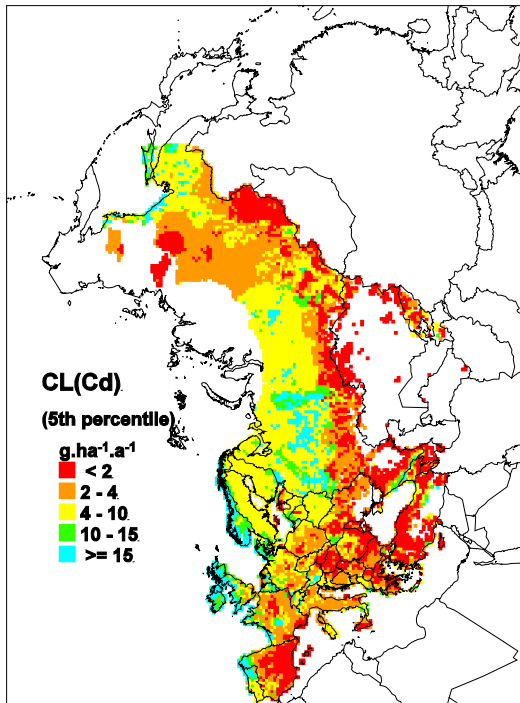


17 countries participated

Effect 1-4: terrestrial
Effect 5: freshwater

Source: RIVM-CCE

Some results: using the CCE background database



Simple critical load approach was also used for Cr, Ni, Cu, Zn, As, Se

Modelled depositions

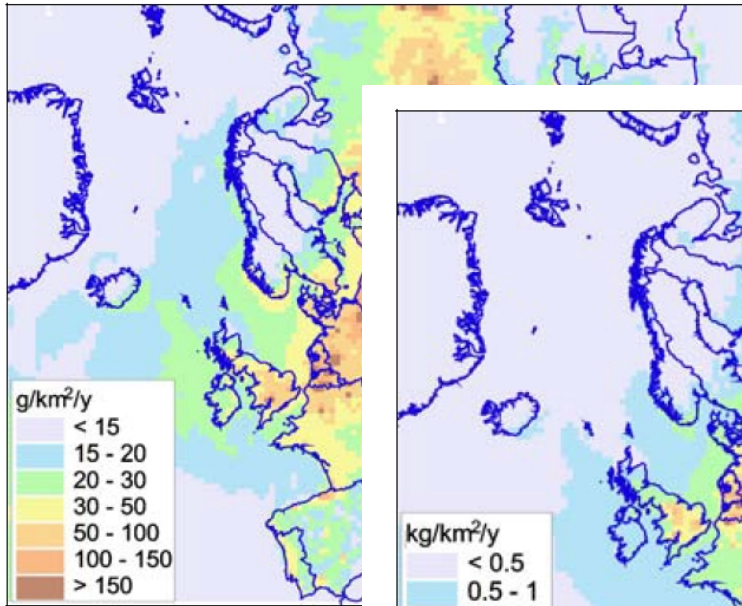


Fig. 27. Spatial distribution of modelled depositions in Europe in 2003

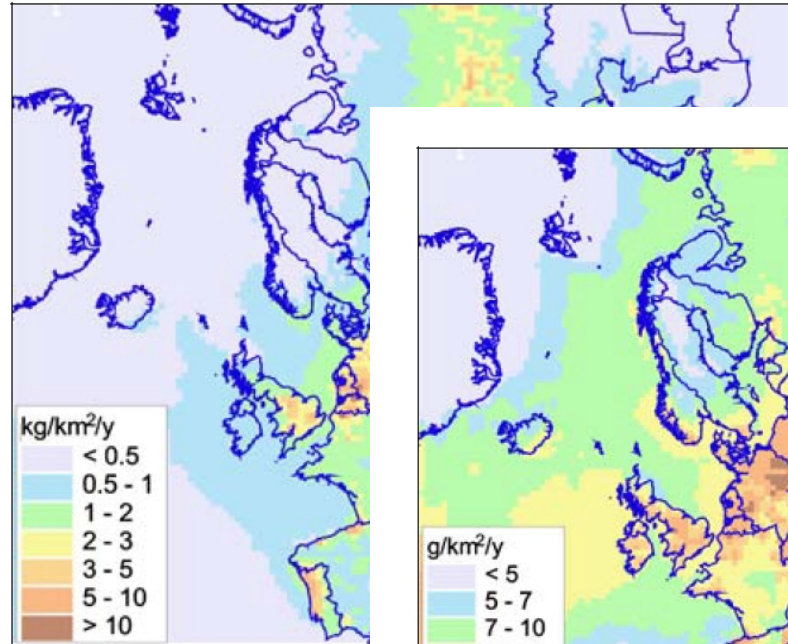


Fig. 22. Spatial distribution of modelled depositions in Europe in 2003

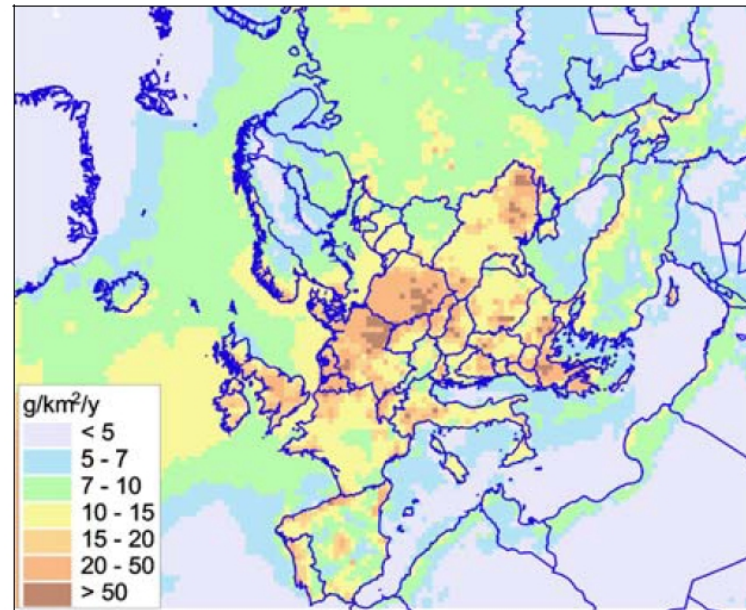
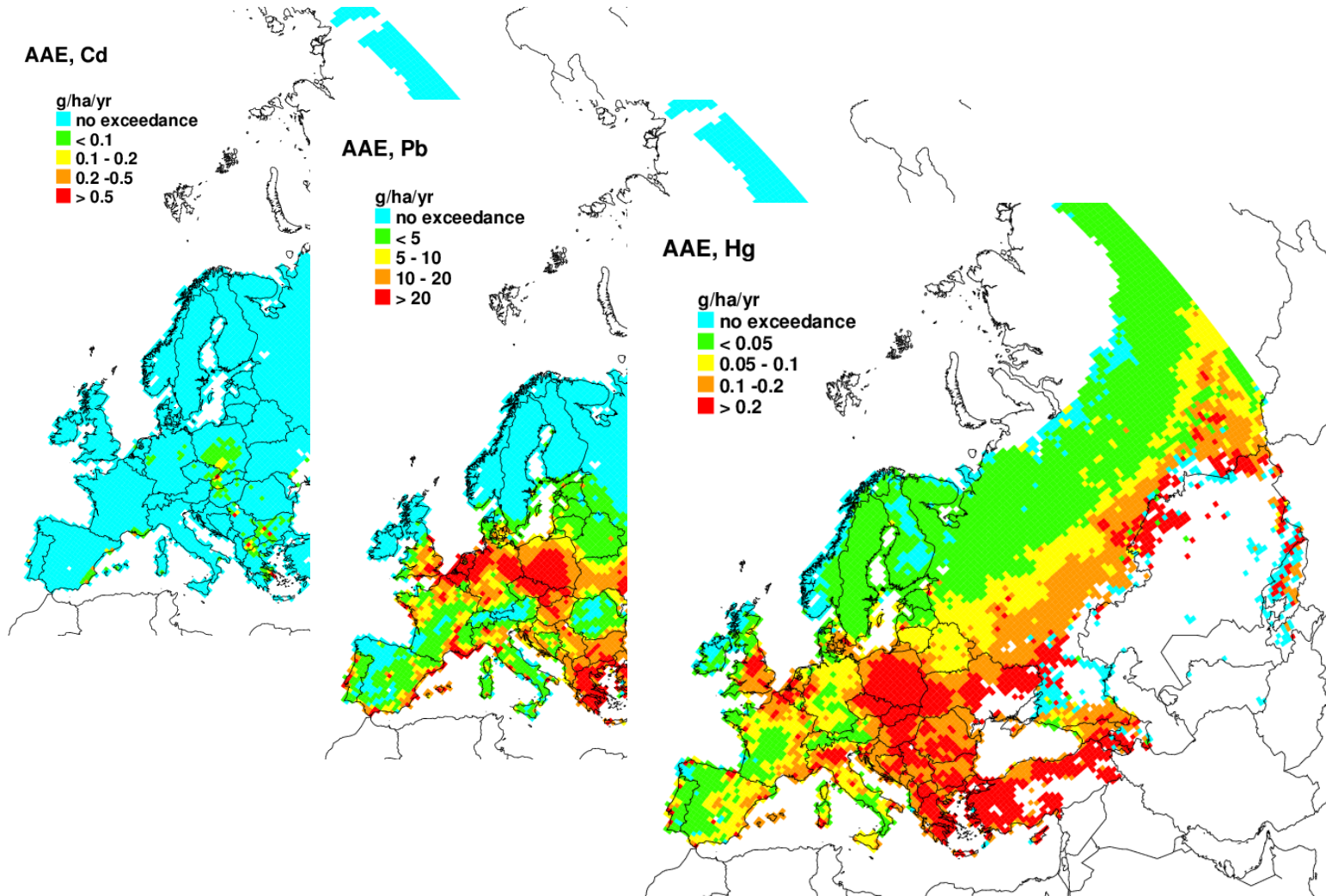


Fig. 32. Spatial distribution of mercury depositions in Europe in 2003

Exceedances (deposition year: 2006)



Conclusions, N + S

- critical loads of N and S have been and are the basis for successful emission reductions
- Substantial reduction of NH_3 emissions remains difficult in Europe; exceedances of CLnutN persist in the future
- Future CL(N) could/will be based on biodiversity endpoints, rather than abiotic limits; this work is under development within EU FPVII Eclairé project

Conclusions, heavy metals

- For heavy metals, critical loads, depositions and exceedances have been computed but have not been used directly in emission abatement
- Other activities (UN Minamata convention) halt the use of CL(HM) for ecosystems, human health is considered more important
- There is a substantial uncertainty in critical limits for HM as well as in emissions and depositions especially for Hg
- Preliminary results show no or very little exceedances for metals other than Cd, Pb and Hg

End

